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## PATENT APPLICATION FEE DETERMINATION RECORD

Substitute for Form PTO-875

Application or Docket Number  
10735494

APPLICATION AS FILED - PART I

(Column 1)

(Column 2)

SMALL ENTITY

OR

OTHER THAN  
SMALL ENTITY

FOR	NUMBER FILED	NUMBER EXTRA
BASIC FEE (37 CFR 1.16(a), (b), or (c))		
SEARCH FEE (37 CFR 1.16(k), (l), or (n))		
EXAMINATION FEE (37 CFR 1.16(o), (p), or (q))		
TOTAL CLAIMS (37 CFR 1.16(j))	minus 20 *	*
INDEPENDENT CLAIMS (37 CFR 1.16(h))	minus 3 *	*
APPLICATION SIZE FEE (37 CFR 1.16(i))	If the specification and drawings exceed 100 sheets of paper, the application size fee due is \$250 (\$125 for small entity) for each additional 50 sheets or fraction thereof. See 35 U.S.C. 41(a)(1)(G) and 37 CFR 1.16(s).	
MULTIPLE DEPENDENT CLAIMS PRESENT (37 CFR 1.16(j))		

DATE	RATE (%)	FEE (\$)
1.50		
25		
100		

RATE (\$)	FEE (\$)
300	
x 50 <sup>c</sup>	
<del>x 200<sup>c</sup></del>	
/	
TOTAL	

\* If this difference in column 1 is less than zero, enter '0' in column 2.

APPLICATION AS AMENDED - PART II

(Columbo 1)

(Column 2)

(C. 2000. 3)

SMALL ENTIRE

684

OTHER THAN  
SMALL ENTITY

AMENDMENT A	CLAIMS REMAINING AFTER AMENDMENT 1	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA
8-18-06 Total: (37 CFR 1.1611)	17	35	:
Independent: (37 CFR 1.1612)	3	4	:
Application Serial: 112-1611			
FIRST PRESENTATION OF MULTIPLE DEPENDENT CLAIM (37 CFR 1.1611)			

RATE (\$)	ADDITIONAL FEE (\$)
.25	
.100	
TOTAL	
DATE	

RATE (\$)	ADDITIONAL FEE (\$)
50	
200	
TOTAL	

15, 19, 28,

AMENDMENT 8	
	Total 12 CFR
	Independent 12 CFR
	Applicable

(Column 1)		(Column 2)	(Column 3)
	CLAIMS REMARKS AFTER AMENDMENT	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT ENTER
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02.11.15	2015 11.11.15 02.11.15

DATE IS	ADD THOMAS FCE IS

1. The enthalpy of fusion of  $\text{C}_{10}\text{H}_8$  is  $10.1 \text{ kJ mol}^{-1}$  and the enthalpy of vaporization of  $\text{C}_{10}\text{H}_8$  is  $33.9 \text{ kJ mol}^{-1}$ .  
 2. The enthalpy of fusion of  $\text{C}_{10}\text{H}_8$  is  $10.1 \text{ kJ mol}^{-1}$  and the enthalpy of vaporization of  $\text{C}_{10}\text{H}_8$  is  $33.9 \text{ kJ mol}^{-1}$ .  
 3. The enthalpy of fusion of  $\text{C}_{10}\text{H}_8$  is  $10.1 \text{ kJ mol}^{-1}$  and the enthalpy of vaporization of  $\text{C}_{10}\text{H}_8$  is  $33.9 \text{ kJ mol}^{-1}$ .

The highest number of groups, 14, is (state independent) of the highest ...

[illegible]